



Geochemical Signatures of Platinum Group Elements in Ultramafic Rocks of Chotanagpur Gneissic Terrain, Eastern India and Their Genetic Control

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Abstract

Ultramafic rocks occur as intrusives in the form of lensoid bodies in the northwestern part of Chotanagpur Gneissic Terrain and consist mainly of olivine, orthopyroxene and clinopyroxene and are devoid of plagioclase. Based on distinct geochemical characteristics, the rocks have been identified as komatiites. The rocks are geochemically analogous to Al-undepleted Munro type (Al₂O₃/TiO₂=17.88-54.73) with distinctly high MgO (26.2-35.62 wt%), Ni (958-1902 ppm) and Cr (21.32-3320 ppm) contents. These rocks are characterised by low CaO/Al₂O₃, (Gd/Yb)n, (La/Yb)n with positive Zr, Hf, Ti anomalies suggesting high degree partial melting of mantle under anhydrous conditions at shallow depth with garnet as a residual phase in the mantle restite. These high MgO volcanic rocks having elevated concentrations of Ni and Cr are potential hosts for Platinum Group Elements (PGE) owing to their primitive mantle origin and eruption at high temperatures. These rocks have low Σ PGE (29-269.02ppb) content with Ir (0.1-0.8ppb) and Ru (1.05-5.78ppb) among Iridium group PGE (IPGE); and Pt (5.04-18.72ppb), Pd(3.5-18.0ppb), Rh (0.22-0.84ppb) among Platinum group PGE (PPGE). The PGE abundances in komatiites were controlled by olivine fractionation. The Major, trace, REE and PGE composition of the rock suggest melting under anhydrous condition at shallow depth above the garnet stability field under S-undersaturated condition. Anhydrous melting associated with mantle plume activity gave rise to the rock which subsequently contaminated by lower crustal materials during magma ascent and emplacement.

Keywords: Chotanagpur Gneissic Terrain, Komatiite, Platinum Group Elements, Sulphur Undersaturation, Plume

Introduction

Platinum Group Elements (PGE: Pt, Pd, Rh, Ir, Os, Ru) show strong chalcophile and siderophile affinity and tradionally they are subdivided into two groups- the compatible IPGEs (Os, Ir, Ru) and the incompatible PPGEs (Rh,Pd,Pt; Fiorentini et al., 2011). The IPGEs often have a refractory characters and are mainly associated with spinel where as PPGEs tend to concentrate in base metal sulphides. PGE compositions in mantle-derived ultramafic-mafic magmas are considered as sensitive indicators to understand their petrogenetic evolution and sulphide saturation history. The behavior of PGEs and their abundance is controlled by several factors like mantle heterogeneity, enrichment-depletion, processes of the mantle, partial mantle melting, melt percolation, sulphide segregation and crystal fractionation (Mondal, 2011 and references therein). Higher concentration of PGE in ultramafic rocks in comparison to other rock types not only suggest their economic significance but their lower concentration are also useful as geochemical tools to evaluate chemical evolution of mantle through time and their role in crust forming processes (Condie and Kroner, 2013). The appearance of Platinum Group of Minerals (PGMs) is rare phenomena and usually develops due to high abundance of

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PGE concentration and favorable sulfur fugacity condition in the magma (Maier, 2005). The PGMs usually occur as disseminated grains in ultramafics and are mainly associated with either sulphides of Ni or Cu or in intergranular spaces between olivine and pyroxene. PGE abundances in mantle derived rocks especially komatiites have been documented by various workers. These rocks reflect degree of partial melting of hot and dry peridotite, sulphur saturation history, rate of ascent, degree of differentiation and contamination prior to eruption which in turn suggest their ability to extract compatible elements and thereby provide information on PGE abundances of mantle (Fiorentini *et al.*, 2011; Said *et al.*, 2011; Zhou *et al.*, 2014; Balaram *et al.*, 2013; Tushipokla and Jayananda, 2013; Manikyamba and Saha, 2014; Saha *et al.*, 2015; Guo *et al.*, 2020; Dora *et al.*, 2022).

Worldwide occurrences of PGE are very limited and confined to plutonic or hypabyssal magmatic intrusions of Archaean-Palaeoproterozoic age. In India three old cratonic blocks *viz* Dharwar in southern India, Singhbhum in Eastern India and Bundelkhand in central India are the Archean terrain where plutonic to hypabyssal ultramafic intrusions of Archaean as well as early Proterzoic age are present. There are three well documented chromite bearing PGE prospects are reported in association with ultramafic complex at (a) Baula Nausahi in Odisha (b) Sittampundi anorthosite complex in Tamil Nadu and (c) Channagiri and Hanumalapura in Karnataka. Recent studies show that there is



Fig. 1. Geological map showing various tectonic elements in the Chotanagpur Gneissic Terrain (Modified after, Rao et al., 2014)

tremendous potential for PGE in mafic-ultramafic and ophiolite complexes occurring in different parts of India.

A number of mafic-ultramafic dykes/sills intrude the basement rocks of Chotanagpur Gneissic Terrain (CGT). In the northwestern part of CGT near Semra village (N23° 59'59": E83° 58'44"), 17 kms southwest of Medninagar town (formely known as Daltonganj town) and adjacent to Gondwana graben, the PGE and associated PGM in the ultramafic rocks of komatiitic composition have been recorded for the first time . A preliminary information on geochemical nature of PGEs content in these rocks has been discussed by Prasad and Bhattacharya (2013). In the present work we reassess major, trace, REE and PGE signatures of these ultramafic rocks to evaluate source characteristic, mantle melting conditions and sulphide saturation history on the basis of newly generated geochemical data.

Geological Setup

A roughly WNW-ESE trending orogenic belt called Central Indian Tectonic Zone (CITZ) dissects the Indian Shield into two halves and represents a major paleosuture along which different Archaean and Palaeoproterozoic blocks of Indian Shield are believed to have fused to form the "Greater Indian Landmass" (Bhowmik, 2019; Negi et al., 2022). The CGT which is a vast tract in the Peninsular Eastern Indian Shield represents the eastern extension of CITZ and comprises of gneisses and migmatites with patches of supracrustals composed of metamorphosed pelites, calcareous and psammitic sediments. Younger mafic, ultramafic and alkaline intrusive are also present. Number of mafic dykes have been studied highlighting the importance of igneous and tectonic processes in CGT (Prasad et al., 2012; Patel et al., 2021). Suites of peridotitic komatiite from Palamau district of Jharkhand (Bhattacharya et al., 2010) indicate that the mantle activities have occurred in the terrain. The ultramafic and mafic rocks of unknown geochemical affinity that emplaced at 1925±110 are suggested to be the oldest magmatic units (Rekha et al., 2011) in CGT.

The rocks of CGT display varying degrees of metamorphism and tectonic deformation occurring at ~2.5, 1.6-2.5, 1.2-1.0 and 0.9 Ga (Sanyal and Sengupta, 2012). Based on published geological and geochronological information a three fold subdivision of CGT is proposed extending from south to north (Mukherjee et al., 2019). The three domains designated as I, II and III are characterized by their gross lithology, metamorphism and deformational history. However, most of the ages reported from CGT are from the metamorphosed basement lithology. Thus the evolutionary history of CGT has remained speculative due to lack of sufficient geological information from the magmatic rocks. Ghose and Chatterjee (2008) recognized five major intra continental rift/ shear zones that control magmatism in CGT. Four of them namely South Purulia Shear Zone (SPSZ), North Purulia Shear Zone (NPSZ), Damodar Graben and South Narmada South Fault (SNSF) located from south to north of CGT trending ENE-WSW to E-W. The fifth runs N-S coinciding with Rajmahal Basin. The present area of investigation lies adjacent to Damodar-Graben (Fig. 1) suggesting emplacement of the studied ultramafic bodies in an extensional tectonic regime.

Analytical Techniques

Major oxides of ten number of representative samples selected on the basis of petrographic study were analysed by XRF at National Geophysical Research Institute (NGRI), Hyderabad (India). A Phillips Magi X PROPW2440 fully automatic, microprocessor controlled, 168-position automatic PW 2540 with VRC sample changer Wavelength dispersive X-ray Spectrometer was used for the determination of major oxides. International geochemical standard reference material from Geological Survey of Japan (JB-2, Basalt) was used to prepare calibration curves for major elements (Krishna *et al.*, 2007). From the analyzed value of JB-2, the error limits of measurement for major elements are within 5%. Trace and rare elements were analysed at NGRI, Hyderabad using Inductively Coupled Plasma Mass Spectrometer (ICP-MS), model Perkin Elmer SCIEX EIAN DRC-II. The analytical procedure, calibration and standard followed are followed *after* Govindaraju (1994). To ensure precision of data, International Standard UB-N (serpentine) was used for calibrating the instrument and closed vessel digestion method has been employed for the analysis. PGE concentrations were estimated by using nickel Sulphide Fire- assay pre-concentration with Te-Co precipitation method followed by Inductively Coupled Plasma Mass Spectrometry (ICP-MS; Perkin Elmer Ellan DRC-II). Reference material WMG-1(Mineralised Gabbro) was utilized for calibrating the instrument and to monitor the accuracy of the data. Au and PGE were analysed following the process suggested by Balaram (2008). The detection limits of most of the elements including PGE were about 0.01mg/ml, and the precision is better than 6% for trace and REE, and is <10% for PGE analysis.

Petrographic Characteristics

The ultramafic rocks are dark grey medium to coarse grained massive and composed mainly of olivine and pyroxene. The primary igneous mineralogy has been altered to serpentinetermolite-actinolite-chlorite-talc-magnetite assemblage. Spinifex zone of this rock exhibits random and platy spinifex textures defined by criss-cross arrangement of primary minerals like olivine and pyroxenes (Fig. 2a). Olivine crystals are altered to serpentine and secondary magnetite whereas pyroxene plates are mostly altered to tremolite. The essential minerals of the cumulate zone are olivine pseudomorphosed by serpentine and magnetite, clinopyroxene and tremolite. These are usually olivine ortho to adcumulates (Fig. 2b) with decreasing amount of intercumulates material towards the base of the flow. Intercumulus phases are mainly pyroxene with fine grained magnetite, Cr-spinel, serpentine and talc. Cumulus olivine grains are dominantly subhedral and equant to tabular. The rock is characterized by complete absence of plagioclase. The large observed variation of LOI content is due to the variable proportions of secondary phases such as serpentine, tremolite, chlorite and talc which attest that the rock have been suffered metamorphism up to greenschist to lower amphibolites facies.

The signature of sulphur enrichment in these rocks is marked by appearance of dark color sulphide phases and sometimes by



Fig. 2 a. Photomicrograph showing platy spinifex texture exhibited by crisscross arrangement of olivine crystals. b. Photomicrograph showing anhedral to subhedral olivine grains randomly oriented making ortho to adcumulates texture

leaching of light yellow colored minerals on the weathered surface. The granular aggregates of PGE bearing minerals and sulphide of nickel and copper are present in the interstitial spaces of cumulates of olivine and pyroxenes.

Results

The results of major oxides, trace elements, REE and PGE analyses of the studied rocks are presented along with analyses of some typical occurrences of the world (Table 1-4).

Table 1: Major oxide (in wt%) analysis of the ultramafic rocks of the study area along with some typical occurrences of the world

	5	(5				2	0	21						
Sample no	PD5	PD11	PD8	PD4	PD3	PD1	PD21	PD6	PD7	PD9	BGB	AGB	GI	WDC	PM
SiO ₂	35.6	36.01	40.23	43.01	39.62	40.66	41.45	36.92	37.01	38.16	47.2	45.1	43.48	45.67	45
TiO ₂	0.2	0.21	0.19	0.22	0.22	0.23	0.41	0.11	0.18	0.22	0.3	0.31	0.49	0.73	0.2
Al_2O_3	5.58	5.22	6.63	7.8	6.57	6.88	7.01	6.02	6.53	6.8	3.4	6.3	9.6	12.61	3.65
Fe ₂ O ₃	9.92	9.68	11.22	10.52	11.02	12.5	12.66	10.08	10.58	10.51		11.4	11.2	9.01	
MnO	0.31	0.12	0.1	0.12	0.15	0.14	0.14	0.22	0.24	0.22	0.2	0.15	0.17	0.11	0.13
MgO	35.62	34.52	31.11	26.2	31.76	30.02	29.15	34.22	33.02	32.88	31.4	30	21.9	21.97	37.8
CaO	5.44	5.61	6.02	6.88	5.98	5.08	5.01	6.52	6.48	6.16	5.6	5.7	8.6	4.18	3.55
Na ₂ O	0.31	0.33	0.28	0.31	0.38	0.07	0.15	0.14	0.27	0.23	0.07	0.34	0.64	0.08	0.33
K ₂ O	0.23	0.27	0.22	0.31	0.02	0.22	0.25	0.37	0.13	0.14	0.04	0.05	0.04	0.05	
P_2O_5	0.03	0.02	0.02	0.03	0.01	0.01	0.02	0.03	0.05	0.02	0.02	0.01	0.04	0.06	
LOI	6.68	7.88	5.01	5.22	5.02	5.21	5.24	4.98	4.21	4.58		7.5	5.3	5.39	
Total	99.74	99.87	101.03	100.62	100.75	101.02	101.79	99.61	100.89	99.9					
Mg#	88.03	87.75	84.61	83.33	84.95	83.14	82.75	87.62	85.05	86.31					
Al ₂ O3/TiO ₂	27.9	24.86	34.89	35.45	29.86	29.91	17.88	54.73	34.61	34					
CaO/Al ₂ O ₃	0.97	1.07	0.91	0.88	0.91	0.78	0.71	1.08	0.99	0.9					
CaO/TiO ₂	27.2	26.71	31.68	31.27	27.18	22.09	12.21	59.27	36	28					
MgO/TiO ₂	178.1	164.4	163.73	119.09	144.36	130.52	71.1	311.09	183.44	149.45					
MgO/Al ₂ O ₃	6.38	6.61	4.69	3.36	4.83	4.36	4.18	5.68	5.05	4.84					

Table 2: Trace elements(in ppm) analysis of the ultramafic rocks of the study area along with some typical occurrences of the world

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Sample no	PD5	PD11	PD8	PD4	PD3	PD1	PD21	PD6	PD7	PD9	BGB	AGB	GI	WDC	PM
Ni	1902	1881	1138	958	1208	1057	1025	1709	1507	1452	1611	1633	997	834.87	1960
Cu	85	81	51	35	66	42	37	78	72	71					30
Co	95	91	60	52	66	63	57	78	71	69					
Cr	3320	3008	2532	2132	2608	2350	2223	2908	2892	2690	2910	2600	2323	2246	2625
Rb	0.49	0.52	0.76	0.81	0.68	0.77	0.83	0.59	0.63	0.63					0.6
Sr	9	11	15	18	14	19	17	11	12	12					19.9
Cs	0.02	0.03	0.04	0.04	0.03	0.05	0.04	0.05	0.04	0.03					0.02
Zn	47	51	44	65	32	55	68	46	40	38					55
Sc	19.2	17.6	25.8	31.2	24.2	33.3	37.7	18.8	22	23.8				17.37	16.2
V	52	82	127	168	108	143	152	76	83	92				153.5	82
Nb	0.11	0.13	0.36	0.88	0.33	0.52	0.82	0.05	0.18	0.27	1.05	0.4	0.6	0.92	0.66
Та	0.6	0.58	0.22	0.28	0.17	0.19	0.22	0.06	0.09	0.09				0.04	0.04
Zr	12	15	33	35	27	36	38	18	21	29	19.1	15	19.2	58.5	10.5
Hf	0.09	0.12	0.37	0.42	0.29	0.4	0.4	0.22	0.22	0.28				1.32	0.28
Y	7.51	7.23	10.81	15	10.9	11.5	13	8.02	8.18	9.52	6.8	7.6	12.3	12	4.3
Ti	1199	1259	1139	1319	1319	1379	2458	659	1079	1319					1205
U	0.02	0.08	0.07	0.44	0.02	0.08	0.03	0.02	0.02	0.08				0.24	0.02
Th	0.02	0.09	0.06	0.45	0.03	0.09	0.02	0.03	0.06	0.09	0.1	0.04	< 0.5	0.49	0.08
Ratios															
Ti/Zr	99.91	83.93	34.51	37.68	48.85	38.3	64.68	36.61	51.38	45.48					
Ti/Y	159.7	174.1	105.36	87.93	121	119.91	189.07	82.16	131.9	138.55					
Ti/Sc	62.44	71.53	44.14	42.27	54.5	41.41	65.19	35.05	49.04	55.42					74.38
Ti/V	23.05	15.35	8.96	7.85	12.21	9.64	16.17	8.67	13	14.33					14.69
Zr/Y	1.59	2.07	3.05	2.33	2.47	3.13	2.92	2.24	2.56	3.04					2.44
Zr/Sc	0.62	0.85	1.27	1.12	1.11	1.08	1	0.95	0.95	1.21					0.65
Sc/Y	2.55	2.43	2.38	2.08	2.22	2.89	2.9	2.34	2.68	2.5					
V/Zr	4.33	5.46	3.84	4.8	4	3.97	4	4.2	3.95	3.17					
V/Sc	2.7	4.65	4.92	5.38	4.46	4.29	4.03	4.04	3.77	3.86					5.06
Ni/Cu	22.37	23.22	22.31	27.37	18.3	25.16	27.7	21.91	20.93	20.45					
Nb/Y	0.01	0.02	0.03	0.06	0.03	0.06	0.06	0.01	0.02	0.03					

The rocks of the area are characterised by low SiO₂ (35.6-43.01), CaO (5.01-6.88), TiO₂ (0.11-0.23) Na₂O (0.07-0.38), K₂O (0.02-0.27), moderate Al₂O₃ (5.22-7.8), Fe₂O₃ (9.68-12.66), high MgO (26.2-35.62) and Loss on ignition (LOI) varies from 4.21 to7.88. Al₂O₃- (Fe₂O₃ + TiO₂)- MgO (Jenson, 1976; Viljoen *et al.*, 1982; Fig. 3a) and CaO- MgO- Al₂O₃ diagrams (Viljoen *et al.*, 1982; Fig. 3b) classify these ultramafic rocks as komatiite which is consistent with their major oxide compositions. Al₂O₃/TiO₂ (17.88-

54.73) and CaO/Al₂O₃ (0.71-1.08) ratios suggest that the komatiites of the area are geochemically analogous to Al-undepleted Munro type komatiites. Al₂O₃, SiO₂, Fe₂O₃, TiO₂ and CaO exhibit negative correlation with MgO (Fig.4a).

The rocks are characterized by high concentrations of compatible trace elements like Ni (958-1902 ppm), Co (52-95 ppm) and Cr (2132-3320 ppm) with relatively lower concentrations of incompatible large ion lithophile elements (LILE) such as Rb (0.49-

Table 3: Rare Earth Elements (in ppm) analysis of the ultramafic rocks of the study area alongwith some typical occurrences of the world

Sample no	PD5	PD11	PD8	PD4	PD3	PD1	PD21	PD6	PD7	PD9	BGB	AGB	GI	WDC	PM
La	0.29	0.46	0.69	0.88	0.66	0.76	0.87	0.56	0.81	0.72	1.2	0.4	0.8	3.25	0.65
Ce	0.87	0.91	1.69	2.31	2	1.93	2.24	0.98	1.44	1.56	3.2	1.3	2.1	9.23	1.6
Pr	0.09	0.17	0.36	0.85	0.17	0.2	0.18	0.1	0.2	0.37				1.33	0.25
Nd	0.51	0.54	1.09	1.81	1.52	1.67	0.81	0.62	0.98	1.31	2.5	1.4	2.1	6.09	1.25
Sm	0.12	0.2	0.41	0.79	0.19	0.43	0.28	0.26	0.28	0.4	0.8	0.6	0.9	1.82	0.41
Eu	0.09	0.09	0.42	0.88	0.42	0.56	0.61	0.11	0.18	0.22	0.3	0.2	0.4	0.53	0.15
Gd	0.56	0.62	0.86	1.15	0.82	0.95	0.88	0.73	0.67	0.57	0.9	0.9	1.5	2.33	0.54
Tb	0.08	0.09	0.15	0.21	0.14	0.17	0.16	0.11	0.11	0.14				0.43	0.09
Dy	18	0.2	0.41	1.32	0.35	0.44	1.03	0.27	0.32	0.32	1.08	1.2	2	2.44	0.67
Но	0.04	0.06	0.21	0.41	0.18	0.27	0.33	0.09	0.11	0.13				0.5	
Er	0.18	0.21	0.67	0.98	0.44	0.72	0.77	0.29	0.38	0.43	0.6	0.8	1.3	1.36	0.44
Tm	0.05	0.08	0.14	0.21	0.13	0.15	0.17	0.11	0.17	0.14				0.21	0.07
Yb	0.68	0.68	1.11	1.33	1.02	1.1	1.12	0.88	0.64	0.66	0.6	0.8	1.3	1.32	0.44
Lu	0.04	0.06	0.17	0.21	0.15	0.18	0.22	0.09	0.12	0.1				0.19	
(La/Nb)n	2.65	3.74	1.93	1.02	2.04	1.48	1.07	10.75	4.63	2.71					
(La/Ta)n	0.03	0.05	0.18	0.18	0.22	0.23	0.23	0.53	0.52	0.46					
(La/Sm)n	0.17	0.16	0.12	0.08	0.25	0.12	0.22	0.15	0.21	0.13					
(Gd/Yb)n	0.66	0.74	0.63	0.71	0.66	0.7	0.64	0.68	0.86	0.71					
(La/Yb)n	0.34	0.46	0.43	0.46	0.44	0.48	0.59	0.44	0.87	0.75					
(Sm/Nd)n	0.24	0.37	0.37	0.44	0.12	0.26	0.35	0.42	0.29	0.31					
(Th/Nb)n	1.49	5.7	1.38	4.3	0.76	1.44	0.2	4.75	2.81	2.78					
(Sm/Yb)n	1.69	2.82	3.53	5.68	1.79	3.75	2.39	2.84	4.2	5.79					
La/Nb	2.63	3.53	1.91	1	2	1.46	1.06	11.2	4.5	2.66					
La/Ta	0.48	0.79	3.13	3.14	3.88	4	3.95	9.33	9	8					
La/Sm	2.14	2.3	1.68	1.11	3.47	1.76	3.1	2.15	2.89	1.8					

Table 4: Platinum Group Eleme	ents(in ppb)analysis of the	ultramafic rocks of the study ar	rea alongwith some typical	occurrences of the world
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Sample no	PD5	PD11	PD8	PD4	PD3	PD1	PD21	PD6	PD7	PD9	BGB	AGB	GI	Chrondrite	PM
Ru	5.78	4.61	1.47	1.05	4.22	1.08	1.29	3.69	3.05	2.88	4.3	5.31	3.3		5
Rh	0.44	0.42	0.78	0.57	0.22	0.71	0.68	0.84	0.81	0.81					0.9
Pd	18	15.55	4.22	4.2	12.01	3.8	3.5	12.78	8.76	6.61	5.51	10.6	12		3.9
Pt	16.62	18.72	6.64	5.23	13.12	6.62	5.04	11.15	9.29	7.13	5.83	11.1	14.1		7.1
Re	10.01	7.6	4.7	2.8	8.8	3	2.2	8.2	5.6	5.8					
Ir	0.8	0.75	0.3	0.1	0.5	0.3	0.11	0.5	0.4	0.1	1.23	2.13	2.05		3.2
Au	62.04	137	238.2	15.05	38.2	34.5	256.2	38.8	14.98	39.12					1
PGE+Au	113.7	184.7	256.31	29	77.07	50.01	269.02	75.96	42.89	62.45					
Cu/Pd	4.72	5.21	12.09	8.33	5.5	11.05	10.57	6.1	8.22	10.74					
Cu/Ir	106.3	108	170	350	132	140	336.36	156	180	710					
Pd/Ir	22.5	20.73	14.07	42	24.02	12.67	31.82	25.56	21.9	66.1				1	1
Pd/Ru														0.79	0.79
Pd/Rh														2.73	2.75
Pd/Pt	1.08	0.83	0.64	0.8	0.91	0.57	0.69	1.15	0.94	0.93				0.53	0.53
Pt/Pd	0.92	1.2	1.57	1.25	1.09	1.74	1.44	0.87	1.06	1.08					1.82
Ni/Pd	105.7	121	269.66	228.09	100.58	278.16	292.86	133.73	172.03	219.67					
Ru/Ir	7.23	6.15	4.9	10.5	8.44	3.6	11.73	7.38	7.63	28.8				1.28	1.27
Pt/Ir	20.78	24.96	22.13	52.3	26.24	22.066	45.82	22.3	23.22	71.3					
Total IPGE	6.58	5.36	1.77	1.15	4.72	1.38	1.4	4.19	3.45	2.98					
Total PPGE	97.1	171.7	249.84	25.05	63.55	45.63	265.42	63.57	33.84	53.67					
IPGE/PPGE	0.07	0.03	0.01	0.05	0.07	0.03	0.01	0.07	0.1	0.06				0.99	0.99
PPGE/IPGE	14.76	32.03	141.15	21.78	13.46	33.07	189.59	12.17	9.81	18.01				1.01	1.01
Present data	: PD1 to I	PD21- ultr	amafic roo	cks of Sen	nra, Palam	au district,	Jharkhan	d							
RGB_Barber	rton Gree	nstone Re	t (Smith a	nd Erlank	1082)										

BGB-Barberton Greenstone Belt (Smith and Erlank, 1982),

AGB(Abitibi Greenstone Belt)- Munro, Alexo, Pyke hill, Boston (Puchtel et al., 2004), GI Gorgona Island Basaltic Volcamism Study Project, 1981, WDC-Western Dharwar Craton (Tushipokla and Jayananda, 2013), PM- Primitive mantle (McDonough and Sun, 1995) and Chondrite (Sun and McDonough, 1989)

Mg#=100Mg/Mg+Fe⁺²

0.83 ppm) and Sr (9-19 ppm). High field strength elements (HFSE) like Nb (0.05- 0.88 ppm),Ta (0.06-0.58) and Hf (0.09-0.4 ppm) show extremely depleted concentrations except Zr (12-38 ppm) that shows relatively higher range. Ni and Cr show positive correlation with MgO whereas HFSE like Zr, Nb, Yb, Sc, Ta as well as Y increase with decreasing MgO content (Fig. 4b). The rock shows



Fig. 3 a. Al₂O₃-(Fe₂O₃+TiO₂)- MgO triangular plot(Jenson, 1976; *modified after*, Viljoen *et al.*, 1982), b. CaO-MgO-Al₂O₃ triangular plot (*after* Vilijoen *et al.*, 1982) showing the komatiitic nature of ultramafic rocks

wide range of Ni and Cu concentrations which is reflected in Ni/Cu ratios. The ratio of Zr/Y is less than 1 and that of primitive mantle value of 2.4. In the primitive mantle normalized multi-element patterns Nb shows negative whereas Ta, Zr, Hf and Ti exhibit positive anomalies (Fig. 5a). LHREEs are 2 to 3 times enriched than HREEs. Gd,Yb, La, Ce, Sm, Nd show increasing trends with decreasing MgO content (Fig. 4c). In the chrondrite normalized spider diagram (Fig. 5b) LREE is depleted to flat to slightly enriched HREE which is consistent with other such komatiites with high Al₂O₃/TiO₂ ratios (Blichert-Toft *et al.*, 2015).

The PGE contents of komatiites are low (<100 ppm) and total PGE ranges from 10.76 to 42.46 ppb with Ir 0.1 to 0.8 ppb, Ru 1.05-5.78 ppb) among IPGE and 5.04-18.72 ppb Pt, 3.5-18 ppb Pd and 0.22-0.84 ppb Rh among PPGE. Os was not taken into account as it is not possible to quantify Os by NiS fire assay method. Au ranges from 14.98 to 256.2 ppb. The concentration of Pd, Pt and Au are higher than the primitive mantle values of 3.9, 7.1 and 1.0 respectively for most samples whereas values of Ru, Ir and Rh are lower than the primitive mantle values (Table 4; McDonough and Sun, 1995). The rock type have Pd/Ir (12.67-66.1) and Pd/Pt (0.57-1.15) higher than the primitive mantle values (Pd/Ir-1.01, Pd/Pt-0.53; McDonough and Sun, 1995). Pt/Pd ratio in most of the samples is more than 1. In most samples Pd/Ir is more than Pt/Pd (Table 4). High ratio of Pt/Ir (20.77-71.3) and enrichment of Pd are attributed to fractional crystallization of parental magma. The Ni/Cu ratios of the rock is very similar to PGE mineralization associated with komatiite magmatism (Lesher and Barnes, 2009). The plots of Ni, Ir, Pt and Pd with MgO show a common positive trend. In the primitive mantle normalized diagram there is sharp anomaly of Pd, Pt and Au. The PGE contents of most of the samples are in the global range of Al-undepleted komatiites with up to 15.1 ppb Pt, 20.9 ppb Pd, 8.99 ppb Ir, 6.35 ppb Ru and 27.7 ppb. Au (Guo et al., 2020). Except for Ir the studied rocks exhibit near similarities with global range.



Fig. 4 a. Geochemical trend of major oxides showing the crystallization trend of ultramafics of the study area (Harker's binary diagram), b. MgO vs trace elements variation diagram for komatiites of the area, c. MgO vs different REEs variation diagram

Discussion

In the northwestern fringe of CGT, ultramafic rocks having mineralogically and chemically similar to komatiite, occur as lenses of varying dimension intercalated with metasedimentary rocks and follow the regional foliation trend (ENE-WSW) of the country rock . The various major oxides viz, Al_2O_3 , SiO_2 , TiO_2 and Fe_2O_3 exhibit negative correlation with MgO which point out a distinct and strong trend of crystallization of the rock caused by olivine accumulation and fractionation. Ni and Cr concentrations have been considered as very sensitive indicators of olivine, spinel and chromite crystallization from ultramafic magma (Maier *et al.*, 2005; Lesher and Barnes, 2009; Naldrett, 2010). Ni and Cr show positive correlation with MgO which could be related to olivine



Fig. 5 a. Primitive mantle - normalized multielement variation diagram of the komatiites of the area (*after*; Sun and McDonough, 1989), b. Chrondrite-REE normalized spider diagram of komatiites of the area (*after* Sun and McDonough, 1989), c. Primitive mantle- normalized Platinum Group Element distribution pattern (McDonough and Sun 1995)

fractionation, HFSE like Zr, Nb, Yb, Ta as well as Y increase with decreasing MgO content consistent with them being incompatible in olivine. Co and Cu are incompatible in most of the silicates and oxide phases that crystallize from mafic-ultramafic magmas but strongly compatible in sulphides in komatiite. Cu abundances increase during fractionation and become concentrated in the magma during removal of other components in crystallizing phases. Co concentration in the rock increases with MgO and are consistent with accumulation of olivine. The abundances of Sc in komatiites are strongly determined by whether all of the garnet was consumed during melting or whether it remained as a residue phase in the residue. The rise in the abundance of Sc with the decreasing MgO content confirms that it is incompatible in olivine. Gd, Yb, La, Ce, Sm, Nd show decreasing values with increasing MgO content consistent with them being incompatible in olivine.

The plot (Gd/Yd)n vs Al_2O_3/TiO_2 (Fig. 6), and ratios of Al_2O_3/TiO_2 , CaO/Al₂O₃, CaO/TiO₂ MgO/TiO₂, MgO/Al₂O₃, (Gd/Yb)n, Nb/Y, La/Nb suggest that the rock of the area is geochemically analogous to Al-undepleted Munro type Komatiite, the most common variety of Mesoarchaean to Neoarchean komatiites globally (Dostal and Muellar, 2004).

Low CaO/Al₂O₃and (Gd/Yb)n and High Al₂O₃/TiO₂ HREE, Y and Zr suggest derivation of the rock from shallow upper mantle without garnet involvement (Maya et al., 2017). In the primitive mantle normalized spider diagram LREE is depleted to flat to slight enriched HREE which is consistent with other such komatiites with high Al₂O₃/ TiO₂ ratios (Blichert and Toft *et al.*, 2015).Low abundance of incompatible lithophile elements (Ti, Nb, Ta, LREE) moderate Zr, Y, MREE, HREE and high abundance of compatible elements (Ni, Ca, Mg) all attest that these rocks are derived from high degree of partial melting of a long term depleted mantle source (Arndt et al., 2008) where Mg rich olivine and orthopyroxene were dominant phase entering the melts. Low large ion lithophile elements and Th/Yb ratio of the studied rocks are consistent with an undepleted mantle source. The undepleted nature of the komatiites are better explained by partial melting in rising mantle plume that originated at depths less than 200 Km where olivine is liquidus phase and magma is less dense than olivine. These rocks are subsequently modified by crustal contamination by lower crustal materials during magma ascent and emplacements. Crustal contamination plays a significant role in sulphur saturation history and brings about PGE mineralization (Dora et al., 2022).

Barnes *et al.*, (2015) suggested that PGE geochemistry can be used to identify the onset of sulphide saturation in magmatic rocks. Partial melting of the primitive mantle lead to the formation of either S-satuaration or S- undersaturated melts (Chen and Xie, 2008). The relationship between Cu and Pd has been identified as an useful indicator of degree of S-saturation of magmas (Chen and Xie, 2008). According to Said *et al.* (2011) Pt, Pd and Cu behave as incompatible elements during fractional crystallization of Sundersaturated magma. High MgO, low Ti, Al undepleted nature of the komatiites of the area exhibit high Cu/Pd ratios lower than that of primitive mantle thereby indicating removal of sulphides during magmatic evolution and S-undersaturation of the magma. The range of Ni/Cu (18.3-27.7) and Pd/Ir (12.67-42) suggest sulphide undersaturation condition for the rock of the present area.

Process Involved in the Formation of PGE

The komatiites of the present area have near chondritic Pt/Pd



Fig. 6. Discrimination diagrams showing geochemical composition of Alundepleted nature of the studied rock (Dostal and Mueller, 2004)

ratios (0.87-1.57), chondrite Pt/Pd=1.9. PGE data on komatiites (>3.3-~2.6 Ga) have suggested that there is trend of increase in Pt and Pd abundance to younger ages in both Al-undepleted and Al-depleted komatiites (Puchtel *et al.*, 2004). The Pt/Pd ratios of (0.87-1.57) may be due to HSE depletion during core formation followed by their addition during the long term event and mixing into Earth's deep mantle (Said *et al.*, 2011).

The rocks have Pd/Ir (12.67-66.1) and Pd/Pt (0.64-1.15) ratios higher than that of primitive mantle (Pd/Ir-1.01; Pd/Pt-0.53) and Chondrite (Pd/Ir-1.0; Pd/Pt-0.53). The Pt/Pd ratio of primitive mantle is 1.82 (McDonough and Sun, 1995) . Fractional crystallization of olivine and chromite during early stages of magmatic differentiation cause decoupling of Pt and Pd resulting into lowering of Pt/Pd ratios (Said et al., 2011) Pd and Ir concentration are controlled by sulphides and olivine respectively (Luguet et al., 2003). In the Pt/Pd ratio komatiites spread around the primitive mantle value of 1. Higher ratios of Pd/Ir with respect of Pt/Pd in the rock may be due to retaining of Ir and Pt within the refractory MSS (Monosulphide Solution) and Fe-Pt alloy whereas Pd would be easily mobilized and removed by Cu-S melt under relatively dry melting condition. Pentlandite dominated interstitial sulphides can show low Ir and high Pd/Ir ratios (Day et al., 2008). Wide range of Pd/Ir suggest that the PGE composition of these rocks are influenced by contamination from continental lithospheric mantle and PGE extraction during sulphide segregation. Maier (2005) have shown that Al-depleted komatiites of 3.5 Ga Barberton have lower PPGE content than Al-undepleted komatiites commonly found in 2.7 Ga greenstone belts and suggested that they formed by moderate degrees of partial melting that left some sulphide in the residue. The Pd/Ir ratios are low at low degree of partial melting and high at high degree of partial melting. When the partial melting reaches up to 20%, sulphides will be consumed which result in increase of Pd content (Said et al., 2011). Based on these observations a high degree partial melting accounts for the lower Pd and Pt contents and Pd/Ir ratios of the studied rock (Pd-3.5-18 ppb, Pt-5.04-18.72 ppb and Pd/Ir-14.07-66.1).

Relative enrichment of PPGE in the studied rock with respect to IPGE reflect fractionation of the two groups. IPGEs like Pt, Ir, Ru show positive correlation with MgO suggesting fractionation of olivine in the parent magma. The studied rock fall in the fields of komatiites and mantle on Ni/Cu vs Pd/Ir discrimination diagram (Chen and Xie, 2008; Fig. 7a). Ni/Pd vs Cu/Ir plot (Barnes *et al.*, 1985; Fig. 7b) shows an olivine fractionation trend for the komatiites of the area suggesting that PGE abundances in these



Fig. 7 a. Pd/Irv s Ni/Cu plot for the studied samples. (Fields *after* Chen and Xie, 2008), b. Cu/Ir vs Ni/Pd variation diagram in which the komatiites samples are following olivine fractionation trend (Fractionation trends *after* Barnes *et al.*, 1985).

rocks were dominantly controlled by olivine fractionation (Naldrett, 2010).

Primitive mantle normalized PGE patterns show (Fig.5c) distinct Pt enrichment which can be described as to early stage of fractional crystallization with >25% MgO. Variation in major elements with respect to Pd/Ir suggest the komatiites have consistently lower range of Pd/Ir with high MgO, low CaO and Al₂O₃ content. This observation indicates fractionation of IPGE into silicates phases during fractional cryastallisation. In the present study the S- undersatuarated komatiiites having PGE contents characterized by 3.5-18.0 ppb Pd and 5.23-18.72 ppb Pt are attributed to 73% partial melting at shallow depth and under anhydrous conditions. Enrichment of Pt, Rh and Pd relative to Iron primitive mantle normalized diagram can be ascribed to higher

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degree of partial melting of upper mantle. The PGE contents of most of the samples are in the global range of Al-undepleted komatiites with upto 15.1 ppb Pt, 20.9 ppb Pd, 8.99 ppb Ir, 6.35 ppb Ru and 27.7 ppb. Au (Guo *et al.*, 2020). Except for Ir the studied rocks exhibit near similarities with global range. High Pt+Pd/Ir+Ru suggest that the rocks are formed by partial melting and crystal fractionation of olivine rich magma (Barnes *et al.*, 1985).

Conclusions:

The ultramafic rocks are characterized by high MgO and chemically similar to komatiites and correspond to Al-undepleted Munro type komatiites suggesting presence of Meso to Neoarchaean segment in CGT. The Major, trace, REE and PGE composition of the rock suggest melting under anhydrous condition at shallow depth above the garnet stability field under Sundersaturated condition. Anhydrous melting associated with mantle plume activity gave rise to the komatiites. The rock shows contamination of lower crustal material during ascent and emplacement of magma. Crustal contamination played a significant role in PGE mineralization. They display primitive mantlenormalized patterns with positive Pt anomalies and is an example of silica hosted Pt-Au type mineralization. Magmatic Ni- Cu- PGE mineralization in the study area may be one of the prospective exploration targets.

Authors' Contributions

Jugnu Prasad: Investigation, Conceptualization, Formal analysis, Writing- original draft. **Deepak K. Bhattacharya:** Supervision, Reviewing and Editing.

Conflict of Interest

There is no conflict of Interest in this research publication.

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